



Pyridine-3-carboxamides as novel CB₂ agonists for analgesia

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4-Cyclopropyl-6-[(2,4-dichlorophenyl)amino]-N-[(tetrahydro-2H-pyran-4-yl)methyl]-3-pyridinecarboxamide

ABSTRACT

We describe herein the medicinal chemistry approach which led to the discovery of a novel pyridine-3-carboxamide series of CB₂ receptor agonists. The SAR of this new template was evaluated and culminated in the identification of analogue **1a** which demonstrated efficacy in an in vivo model of inflammatory pain.

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Cannabis sativa has been used as an analgesic as early as 2800 BC in China.¹ Cannabinoids are the pharmacologically² active components of cannabis, and are known to mediate some of their actions through the cannabinoid receptors³ CB₁ and CB₂, both of which are G-protein coupled receptors. The CB₁ receptor is found in the central nervous system (CNS),⁴ as well as in the periphery,⁵ whereas the CB₂ receptor is found mainly in the periphery,⁶ particularly in the immune system.⁷ There is also increasing recent evidence of CB₂ receptor expression in the CNS.⁸ Whilst stimulation of the CB₁ receptor mediates analgesia,⁹ this receptor also mediates other undesirable CNS side-effects.¹⁰ CB₂ agonists, on the other hand, have shown efficacy in preclinical models of inflammatory and neuropathic pain.¹¹

We have therefore sought to design selective CB₂ agonists to avoid the undesirable psychoactive effects and abuse potential associated with agonism of the CB₁ receptor. Recently our research into a

novel class of selective CB₂ agonists culminated in the discovery of **1** (GW842166X, Fig. 1),¹² which has entered human clinical studies for the treatment of pain. In this letter we describe our attempts to de-

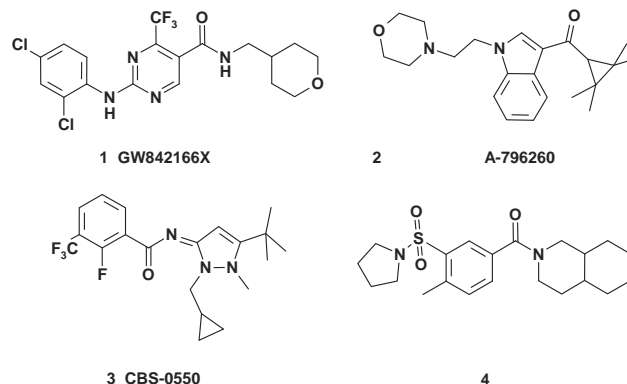


Figure 1. Recent lead CB₂ agonists.

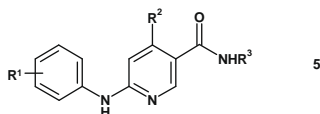
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sign a new series of selective CB₂ agonists with improved aqueous solubility over GW842166X (solubility at pH 7.4 is 2 µg/mL).

Interest in selective CB₂ agonists has been growing, and, in addition to **1**, a number of other groups have reported novel selective CB₂ agonists. Abbott¹³ have disclosed the morpholine **2**, A-796260, whilst Taisho¹⁴ have revealed their lead compound, an iminopyrazole **3**, CBS-0550, and Boehringer Ingelheim-Evotec¹⁵ have published a novel series of selective CB₂ agonists, exemplified by the sulphonamide **4**.

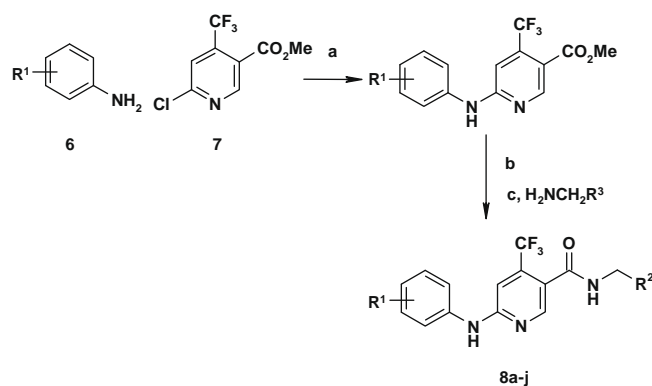
Following the identification of **1** as a clinical candidate, we began to look for a CB₂ agonist back-up candidate with improved aqueous solubility over **1**. As part of this programme, we investigated replacing the central pyrimidine core with a pyridine core, since the basic pK_a of pyridines (5.2) is higher¹⁶ than pyrimidines (1.3), and molecular overlays suggested good alignment¹⁷ of key features of the pyridine **5** with **1**. In the current research presented here, we describe the structure–activity–relationships (SAR), the pharmacokinetics, and efficacy in a pain model of these novel pyridine CB₂ agonists.



Initially we chose to prepare the 4-(trifluoromethyl)pyridines **8a–j**, which retained the 4-(trifluoromethyl)-substituent, so that the direct effects of changing the pyrimidine core to a pyridine core could be compared with our earlier SAR evaluation which led to the discovery of **1**. Condensation¹⁸ of the anilines **6** with the chloropyridine **7**, followed by hydrolysis and amide formation gave the final products **8a–j** (Scheme 1).

All but one (**8j**) of the analogues **8a–j** were found to be CB₂ agonists with an efficacy not significantly different from the reference CB₂ agonist HU210 (Table 1—the assay for CB₂ was run using a yeast CB₂ reporter assay in yeast cells, *Saccharomyces cerevisiae*, expressing the human CB₂ receptor¹⁸). Initially the amide nitrogen substituent was kept fixed as a (4-tetrahydropyranyl)methyl group, **8a–e**, whilst the aniline substituent R¹ was varied. This revealed that a 2,4-dichloro-substituent, as in **8a**, was optimal for CB₂ potency. However, variation of the amide N-substituent was well tolerated and other cycloalkyl groups, **8f–h**, gave CB₂ agonists of comparable potency and efficacy to the tetrahydropyranyl group. Indeed, the cyclohexyl group improved CB₂ potency, (**8h** cf **8b**). A 4-fluorobenzyl group **8i** was similarly tolerated on the amide nitrogen, but the 4-carboxamidobenzyl group in **8j** was not.

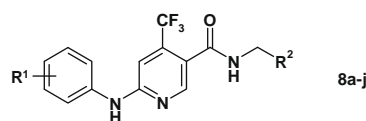
All of the analogues showed selectivity over CB₁ with a pEC₅₀ of less than 5.5 and/or an efficacy of less than 10% at CB₁.



Scheme 1. Reagents and conditions: (a) Neat, 120 °C; (b) KOH, H₂O, reflux; (c) EDC, HOBT or HATU, PS-carbodiimide, HOBT or PS-diisopropylethylamine, 1-butyl-3-methylimidazolium hexafluorophosphate.

Table 1

CB₂ Agonist potencies¹⁸ and efficacies for compounds (**8a–j**).



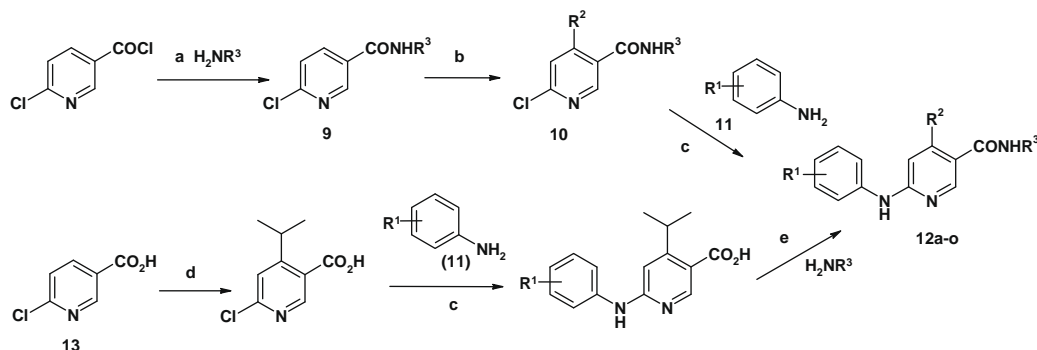
Compound	R ¹	R ²	pEC ₅₀ ^a	SD ^b	E % ^c
8a	2,4-di-Cl	4-Tetrahydropyranyl	7.3	0.3	100
8b	3-Cl	4-Tetrahydropyranyl	6.6	0.3	95
8c	3-F	4-Tetrahydropyranyl	6.0	0.1	90
8d	4-Cl,2-dMe	4-Tetrahydropyranyl	6.6	0.1	94
8e	2-MeO,5-Cl	4-Tetrahydropyranyl	6.4	0.2	89
8f	2-MeO,5-Cl	Cyclobutyl	6.7	0.05	87
8g	2,4-di-Cl	Cyclopentyl	7.0	0.2	86
8h	3-Cl	Cyclohexyl	7.5	0.3	84
8i	3-Br	4-Fluorophenyl	7.1	0.05	84
8j	3-Cl	4-H ₂ NCO-Phenyl	5.8	0.1	10

^a CB₂ assay data is the mean of at least two determinations, see Ref. 18 for the method, HU210 had pEC₅₀ 8.6 SD 0.2 and was used as a reference.

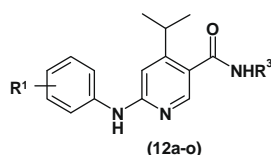
^b Standard deviation.

^c Efficacy (E) is expressed as a percentage relative to the efficacy of HU210 (an unselective CB₁/CB₂ agonist. All analogues show little or no activity at the human CB₁ receptor, that is, CB₁ pEC₅₀ < 5.5 and/or E < 10% (data not shown).

The aqueous solubility¹⁹ of the 4-(trifluoromethyl)pyridines **8a–j** was investigated, but it was not improved compared to **1**: for example, **8a** has a solubility of <1 µg/mL at pH 7.4. We therefore sought to improve the solubility by synthesizing¹⁸ the 4-(isopropyl)pyridines **12a–o** (R² = 'Pr, Scheme 2, Table 2), as the pK_a of 4-(isopropyl)pyridine²⁰ is 6.02 compared to the pK_a of 2.63 for 4-(trifluoromethyl)pyridine.²¹



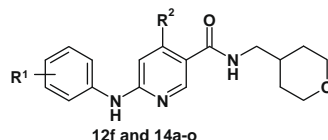
Scheme 2. Reagents and conditions: (a) Et₃N, 0 °C; (b) R²MgBr, 0 °C → rt, then DDQ; (c) microwave 150–180 °C or microwave, MeOH, 180 °C; (d) ^tPrMgBr, 0 °C → rt, then Mn(OAc)₃ –60 °C → rt; microwave, 120 °C; (e) EDC, HOBT.

Table 2
CB₂ agonist potencies¹⁸ and efficacies for compounds **12a–o**

Compound	R ¹	R ^{3a}	pEC ₅₀ ^b	SD ^b	E % ^b
12a	3-Cl	(4-THP)methyl-	6.9	0.15	101
12b	3-Br	(4-THP)methyl-	7.2	0.1	105
12c	3-MeO-	(4-THP)methyl-	6.4	0.1	82
12d	3-Me-	(4-THP)methyl-	6.3	0.2	88
12e	3-NC-	(4-THP)methyl-	6.4	0.1	100
12f	2,4-di-Cl-	(4-THP)methyl-	7.3	0.04	80
12g	3-Cl, 4-NC-	(4-THP)methyl-	7.1	0.15	75
12h	2-F,3-CF ₃ -	(4-THP)methyl-	7.2	0.1	98
12i	3-Cl	Cyclopentylmethyl-	7.4	0.15	105
12j	3-Me	Cyclobutylmethyl-	7.3	0.2	87
12k	3-Cl	4-Fluorobenzyl-	7.0	0.1	89
12l	3-Cl	4-THP-	6.3	0.2	85
12m	3-Cl	ⁱ PrCH ₂ -	7.4	0.1	121
12n	3-Cl	^t BuCH ₂ -	6.6	0.4	97
12o	3-Cl	Et ₂ CHCH ₂ -	6.4	0.25	86

^a THP, tetrahydropyranyl.^b See footnotes in Table 1 for definition of these terms. All analogues show little or no activity at the human CB₁ receptor, that is, CB₁ pEC₅₀ < 5.5 and/or E ≤ 7% (data not shown).

The 4-(isopropyl)pyridines **12a–o** (R² = ⁱPr) were prepared¹⁸ by addition of a Grignard reagent to the pyridinecarboxamide **9**, followed by reaction of **10** with an aniline **11**. Alternatively, addition of isopropylmagnesium chloride to the nicotinic acid **13**, followed by reaction with an aniline **11**, and standard amide formation also gave the desired 4-(isopropyl)pyridines **12a–o** (R² = ⁱPr).

Table 3
CB₂ potencies, in vitro and in vivo pharmacokinetics, and aqueous solubility

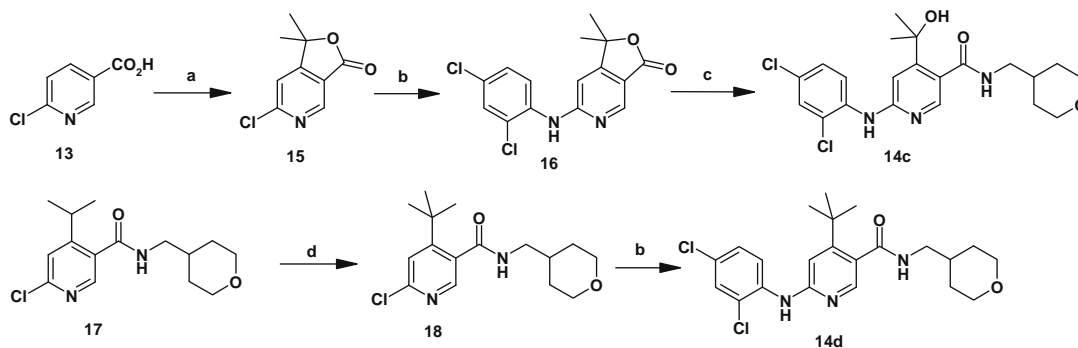
Compound	R ¹	R ²	CB ₂			rCLi ^b	hCLi ^b	CYP450 IC ₅₀ (μM)					AUC (0 – t)/dose (min kg/L) ^c	Solubility μg/mL at pH 7.4
			pEC ₅₀ ^a	SD ^a	E ^a			1A2	2C9	2C19	2D6	3A4		
12f	2,4-dichloro	ⁱ Pr	7.3	0.04	80	7.6	4.0	60	40	43	28	58	nd	5
14a	2,4-dichloro	Cyclopropyl	7.1	0.2	90	4.3	1.7	nd	43	82 ^e	55	>100	15	7
14b	2,4-dichloro	Cyclopentyl	6.8	0.1	61	nd	nd	nd	nd	nd	nd	nd	nd	<1
14c	2,4-dichloro	–C(OH)Me ₂	5.9	0.05	79	nd	nd	nd	nd	nd	nd	nd	nd	150
14d	2,4-dichloro	^t Bu	7.9	0.1	100	2.3	<0.5	22	17	28	>100	64	16	23
14e	3-Cl	Cyclopropyl	6.8	0.1	82	nd	nd	nd	nd	nd	nd	nd	nd	91
14f	3-Br	Cyclopropyl	6.7	0.1	93	1.6	<0.5	23	24	29	16	>100 ^e	5	45
14g	3-CF ₃	Cyclopropyl	6.7	0.3	76	nd	nd	nd	nd	nd	nd	nd	nd	19
14h	3-OCF ₃	Cyclopropyl	6.6	0.1	87	3.5	1.0	NT	16	65	24	57 ^d	3	<1
14i	3-NC	Cyclopropyl	6.1	0.2	78	nd	nd	nd	nd	nd	nd	nd	nd	4
14j	3-MeO	Cyclopropyl	5.7	0.3	65	nd	nd	nd	nd	nd	nd	nd	nd	168
14k	3-Me	Cyclopropyl	5.8	0.2	76	nd	nd	nd	nd	nd	nd	nd	nd	26
14l	3-F	Cyclopropyl	5.6	0.1	87	nd	nd	nd	nd	nd	nd	nd	nd	125
14m	2,5-dichloro	Cyclopropyl	6.4	0.1	92	12.0	2.4	53	27	49	42	>100	nd	12
14n	3,4-dichloro	Cyclopropyl	6.8	0.1	70	nd	nd	nd	nd	nd	nd	nd	nd	6
14o	3,5-dichloro	Cyclopropyl	7.2	0.4	57	nd	nd	nd	nd	nd	nd	nd	nd	7

^a See footnotes in Table 1 for definition of these terms. All analogues show little or no activity at the human CB₁ receptor, that is, CB₁ pEC₅₀ < 5.5 and/or E ≤ 5% (data not shown).^b In vitro clearance, mL min^{–1} g^{–1} liver; r, rat; h, human; see Ref. 22 for procedures.^c Administered to rats as a suspension in aqueous 1% methylcellulose.^d Interference in the assay. nd, not determined.

The in vitro data (Table 2) showed that these 4-isopropylpyridines **12a–o** were potent CB₂ agonists with an efficacy not significantly different from the reference CB₂ agonist HU210. Exploration of the aniline substituent R¹ to investigate the electronic requirements showed that the optimal 3-substituent for CB₂ potency was a halogen, either chlorine or bromine **12a,b**. However, di-substitution was tolerated, **12f–h**, and yielded analogues of similar potency to **12a**. Modification of the amide N-substituent showed that the cyclobutyl ring increased the CB₂ potency, **12j** cf **12d**, and that a number of other rings, but not all (e.g., **12l**), were tolerated **12i–k**. The isobutyl group gave a potent CB₂ agonist, **12m**, with a supra-maximal response. Other N-alkyl substituents were less potent **12n,o**. All analogues in this series were selective over CB₁, with CB₁ pEC₅₀ ≤ 5.5 and/or an efficacy of <7%.

However, the solubility of the 4-(isopropyl)pyridine series, despite a small improvement, was still unacceptably low, for example, **12f** has a solubility of 5 μg/mL at pH 7.4. In addition, preliminary in vitro rat clearance²² for many of the more potent analogues, where measured, was also unacceptably high (e.g., for **12b,f,i,m**, rat CLi > 5 mL min^{–1} g^{–1} liver), and so further optimization was undertaken.

Our attention turned to investigate the SAR of the 4-substituent, that is, R², (Table 3). Analogues with an R² cycloalkyl group **14a,b** were prepared²³ by the routes previously used (Scheme 2), however, the introduction of a 4-(2-hydroxyprop-2-yl)-group or a 4-(*tert*-butyl) group required new syntheses (Scheme 3). Ortholithiation of the acid **13**, followed by treatment with acetone gave the lactone **15**. Displacement of the chlorine with an aniline gave the precursor **16**. Reaction²³ of **16** with an amine using the conditions described by Weinreb,²⁴ effected ring opening to give the 4-(2-hydroxyprop-2-yl)-analogue **14c**. The isopropyl intermediate **17**, (prepared by the routes shown in Scheme 2) was alkylated^{25a} to give the *tert*-butylpyridine **18**, and this converted by the usual methods into the final *tert*-butyl analogue **14d**.



Scheme 3. Reagents and conditions: (a) LiTMP, -55°C , then Me_2CO , -71°C \rightarrow rt; (b) 2,4-dichloroaniline, MsOH, microwave, 180°C ; (c) (4-THP)methylamine (5 equiv), Me_3Al (5 equiv), CH_2Cl_2 , 40°C ; (d) *n*-BuLi (2 equiv) -65°C , MeI.

Modification of the R^2 substituent had a significant effect on the CB_2 potency of these analogues (Table 3). A cyclopropyl group in **14a** maintained the same level of CB_2 potency as an isopropyl group in **12f**, and the *tert*-butyl group in **14d** increased CB_2 potency. However, the cyclopentyl group in **14b** reduced CB_2 efficacy and the (2-hydroxyprop-2-yl) group in **14c** reduced CB_2 potency.

Following preliminary *in vitro* pharmacokinetic analysis, we noticed that the rat *in vitro* clearance²² for the cyclopropylpyridine **14a** was significantly lower than that of the isopropylpyridine **12f**. A final round of optimization of this 4-(cyclopropyl)pyridine sub-series was therefore undertaken prior to evaluation *in vivo* to ensure the best chance of a compound with high oral bioavailability and also activity *in vivo*.

Further 4-(cyclopropyl)pyridine analogues **14e–o** (Table 3) were prepared^{25b} as shown in Scheme 2. Amongst the substituents investigated at the 3-position of the aniline, the halogen containing substituents appeared to give the more potent analogues **14e–h** but 2,4- and 3,5-dichloro-substitution, gave the most potent CB_2 agonists **14a,o**.

The solubility data for this series revealed that two of the oxygen-containing analogues, **14c** and **14j**, as expected, had the greatest aqueous solubility at pH 7.4. Some of the less lipophilic analogues, for example, **14i**, also had improved solubility. However these three analogues also had weaker CB_2 potency and were thus not progressed any further.

The *in vitro* clearance and inhibition of cytochrome P450 for 4-(cyclopropyl)pyridines with CB_2 $\text{pEC}_{50} \geq 6.4$ and $E > 85\%$ was further assessed to prioritise them for *in vivo* pharmacokinetic analysis. Gratifyingly, a number of these, **14a,d,f,h**, appeared to have low *in vitro* clearance in both rat and human microsomes, combined with no significant P450 interactions.

These four analogues were next evaluated in a pharmacokinetic study in conscious rats, to assess their exposure in the blood up to 6 h after an oral dose of 3 mg/kg. Two of these, **14a,d**, had good oral exposure [as expressed by dose normalised area under the curve, $\text{AUC}(0-t)/\text{dose}$] despite having low aqueous solubility.

Since **14a** showed no activity at CB_1 at concentrations up to $30\text{ }\mu\text{M}$ (cf **14d**²⁶), further pharmacokinetic studies were undertaken. In conscious rats, following an oral dose of 3 mg/kg of **14a**, the bioavailability was 39%, whilst an *iv* dose of 1 mg/kg of **14a** produced a clearance of 14 mL/min/kg, a volume of distribution of 2.4 L/kg, and a half life of 3.1 h.

14a has high *in vitro* CB_2 potency and efficacy, with excellent selectivity over CB_1 , and good *in vitro* and *in vivo* pharmacokinetic properties. Despite its low aqueous solubility, it was selected for further evaluation in an *in vivo* model to give encouragement for further research around this template in the future. In the established Complete Freund's Adjuvant (CFA) model of inflammatory pain²⁷, **14a** has high potency with an oral ED_{50} of 0.07 mg/kg *po*,

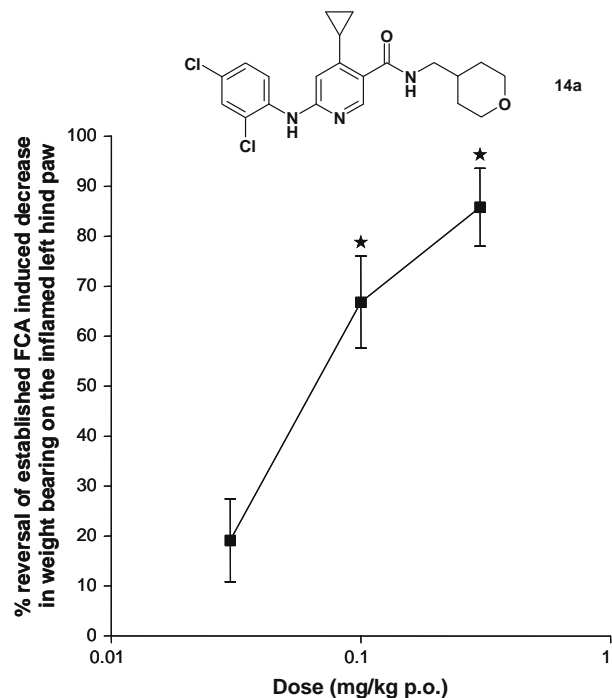


Figure 2. CFA model of inflammatory pain.

and shows full reversal of hyperalgesia at a dose of 0.3 mg/kg *po* as determined by the weight bearing protocol (Fig. 2).

In conclusion, a novel series of potent, selective pyridine-3-carboxamide CB_2 agonists has been discovered. One of these, **14a**, has good oral bioavailability, low clearance, and potent oral activity in the CFA model of inflammatory pain. Whilst some pyridine analogues have improved aqueous solubility, the lower solubility of **14a** is not a barrier to good oral bioavailability and oral activity in the CFA model of inflammatory pain.

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19. The solubility was determined in a high-throughput solubility assay, using the principle of precipitative solubility. A 10 mM DMSO stock solution of the analyte was diluted down in phosphate-buffered saline at pH 7.4. The resulting solution was equilibrated at room temperature for 1 h and filtered to remove any undissolved material. The solubility of the analyte in the filtrate was determined by comparison to a solution of known concentration using hplc/uv detection. True equilibrium is not guaranteed.
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